

A CONVENIENT PREPARATION OF MALTOL, ETHYLMALTOL, AND PYROMECONIC ACID FROM

2-ALKYL-6-METHOXY-2H-PYRAN-3(6H)-ONES

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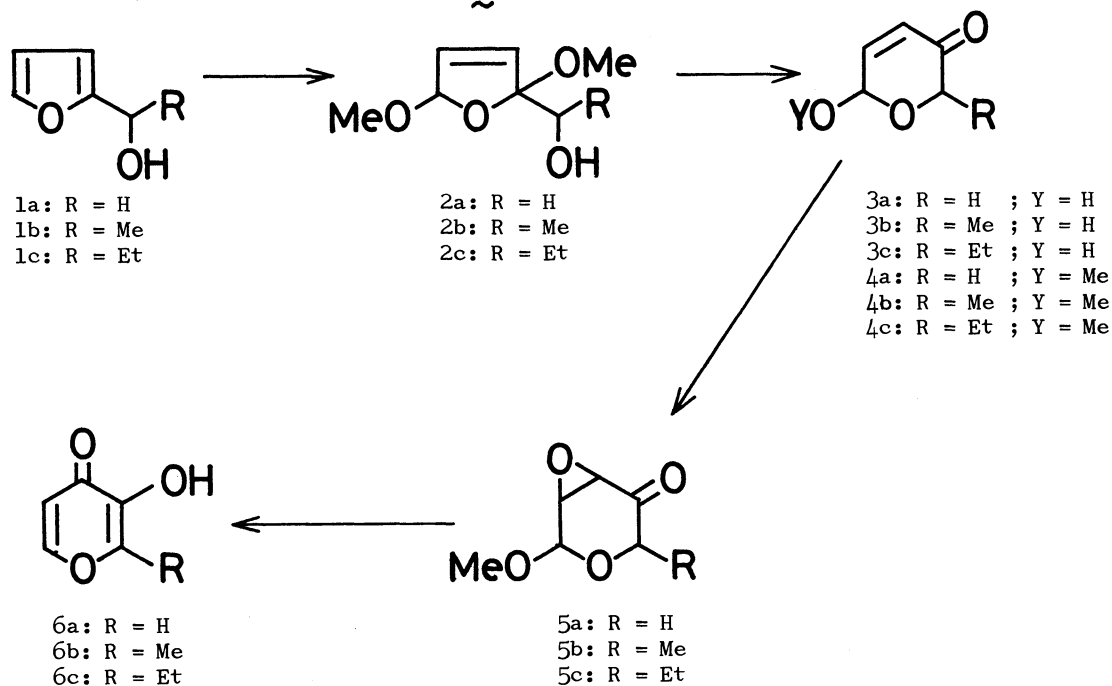
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The efficient synthesis of 2-alkyl-3-hydroxy-4H-pyran-4-ones, e. g., pyromeconic acid (6a), maltol (6b), and ethyl maltol (6c), from 2-alkyl-4,5-epoxy-6-methoxytetrahydropyran-3-ones 5 by acid-catalyzed rearrangement is described. The key precursors 5 were obtained by epoxidation of 2-alkyl-6-methoxy-2H-pyran-3(6H)-ones 4, which were synthesized by three steps starting from furfuryl alcohols 1.

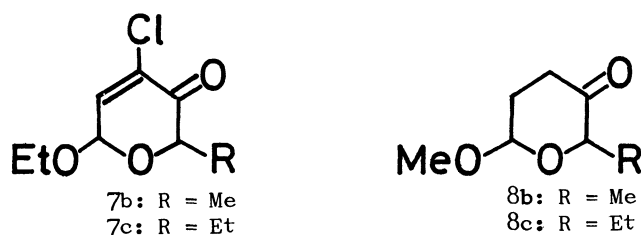
Maltol (3-hydroxy-2-methyl-4H-pyran-4-one) (6b) and its related compounds 6a and 6c have been of great interest as the flavoring additives in foods.¹⁾ For the synthetic purpose of 3-hydroxy-4H-pyran-4-ones 6, kojic acid,²⁾ meconic acid,³⁾ comenic acid,⁴⁾ pyromeconic acid,⁵⁾ rhamnopyranoside,⁶⁾ 1,1-dimethoxy-3,5-heptanedione,⁷⁾ and the related compounds⁸⁾ have been frequently employed as a starting material. Strange enough, attempt to prepare 6 from 2H-pyran-3(6H)-ones 3, which can be easily provided by ring expansion of 2,5-dialkoxy-2-(1-hydroxyalkyl)-2,5-dihydrofurans 2,⁹⁾ has not yet been reported. We wish to report here a new and efficient synthetic way to 2-alkyl-3-hydroxy-4H-pyran-4-ones 6 via 2H-pyran-3(6H)-ones 4 and tetrahydropyran-3-ones 8, which were obtained by several steps starting from furfuryl alcohols 1.

Ring expansion of 2-(1-hydroxypropyl)-2,5-dimethoxy-2,5-dihydrofuran (2c) in aqueous 50% THF solution containing a few drops of 1 M aqueous perchloric acid¹⁰⁾ at room temperature for 10 min afforded 2-ethyl-6-hydroxy-2H-pyran-3(6H)-one (3c) in 98% yield; IR 3330 (OH), 1675 (C=O), 1632 cm⁻¹ (C=C). The preparation of the key intermediate 2c from 1c was carried out using electrochemical procedures improved in our laboratory.¹¹⁾ Refluxing the dihydropyran-3-one 3c with methyl orthoformate in the presence of magnesium sulfate¹²⁾ gave the acetal 4c in 90% yield; IR 1702 (C=O), 1686 (C=O), 1631 cm⁻¹ (C=C); PMR δ 3.47, 3.51 (2s, 3H, CH₃O)(see Table 1). Addition of 2 molar amounts of 15% H₂O₂ to a two-phase solution

containing 4c in diethyl ether and aqueous 5% sodium carbonate at 5-10 ° C provided the epoxy ketone 5c in 88% yield; IR 1726 cm^{-1} (C=O); PMR δ 3.33-3.75 (m, 2H, H-C-O-C-H) (see Table 2). Heating of 5c in dioxane in the presence of sulfuric acid afforded 6c in 91% yield (see Table 3). On the other hand, reaction of the epoxide 5c with 1:20 concentrated hydrochloric acid-ethanol



at 75-80 ° C for 20 hr afforded the chlorinated product 7c¹³⁾ (34%); IR 1710 (C=O), 1627 cm^{-1} (C=C); PMR δ 6.98 (d, 1H, HC=C) along with 6c (39%). Refluxing 5c with 1:4:16 hydrochloric acid-water-ethanol gave exclusively 6c in 67% yield.



The alternative way to 6 was accomplished via tetrahydropyran-3-ones 8. Thus, hydrogenation¹⁴⁾ of 4c in methanol in the presence of PtO_2 gave 8c in 87% yield; IR 1730 cm^{-1} (C=O). Refluxing 8c with selenium dioxide in aq. 60% ethanol afforded 6c in 58% yield.

Studies on the development of the further alternative synthetic ways for maltol synthesis from furfuryl alcohols 1 are currently in progress.

Table 1. Conditions and Results in Acetalization of 3

Compound (g)	CH(OMe) ₃ ml	MgSO ₄ g	MeOH ml	Time h	Bp ° C/ mmHg	Product	Yield %
<u>3a</u> (1.14)	15	3	0.3	6	68-70/11	<u>4a</u> ^{9a)}	77
<u>3b</u> (1.28)	20	3	---	6	72-74/11	<u>4b</u> ^{9a)}	87
<u>3c</u> (1.42)	25	3	---	20	74/10	<u>4c</u> ¹⁶⁾	90

Table 2. Conditions and Results in Epoxidation of 4

Compound (g)	Solvent (ml)	5% Na ₂ CO ₃ ml	15% H ₂ O ₂ g	Temp ° C	Time h	Bp ° C/ mmHg	Product	Yield %
<u>4a</u> (2.56)	Ether (40)	10	9.2	5	3	75-78/11	<u>5a</u> ¹⁵⁾	89
<u>4b</u> (2.84)	Ether (40)	10	9.2	10	3	83-86/11	<u>5b</u> ¹³⁾	90
<u>4b</u> (2.84)	MeOH (30)	10	4.6	0-3	3	83-86/11	<u>5b</u> ¹³⁾	93
<u>4b</u> (2.84)	H ₂ O (30)	10	14.5*	0-5	3	83-86/11	<u>5b</u> ¹³⁾	90
<u>4c</u> (3.12)	Ether (40)	10	9.2	5	3	80-85/5	<u>5c</u> ¹⁶⁾	88

* Aqueous 5% hydrogen peroxide was used.

Table 3. Conditions and Results in Isomerization of Epoxides 5*

Compound (g)	Solvent (ml)	Catalyst (g)	Time h	Product	Yield %
<u>5a</u> (2.88)	Dioxane (25)	80% H ₂ SO ₄ (3.5)	1	<u>6a</u> ¹⁷⁾	78
<u>5b</u> (3.16)	H ₂ O (30)	conc H ₂ SO ₄ (3.0)	4	<u>6b</u> ⁵⁾	87
<u>5c</u> (3.44)	Dioxane (25)	80% H ₂ SO ₄ (3.5)	1	<u>6c</u> ⁵⁾	91

* The reactions were carried out at 95-100 ° C.

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- 10) It has been shown that the isomerization of 2c to 3c was carried out on treatment with 1-2% sulfuric acid for ca. 2 hours: Ref. 9.
- 11) Electrolytic methoxylation of furfuryl alcohols (1a and 1b) in MeOH using NH_4Br as a supporting electrolyte have been shown to give 2a (66%) and 2b (73%), respectively: see N. Clauson-Kaas, F. Limborg, and P. Dietrich, Acta Chem. Scand., 6, 545 (1952); P. Nedenskov, N. Elming, J. T. Nielsen, and N. Clauson-Kaas, *ibid.*, 9, 17 (1955). However, the improved method in MeOH containing Et_4NClO_4 under a constant current of 0.07-0.01 A/cm² (2.02 F/mol) gave 2a (91%), 2b (95%), and 2c (94%), respectively. The details will be published elsewhere.
- 12) The acetalization of the hemiacetals (3a and 3b) using $\text{BF}_3\text{-OEt}_2$ or SnCl_4 as catalysts has been shown to give 4a (43%) and 4b (45%), respectively: Ref. 9a.
- 13) The related chlorinated compound of 7b has been obtained by the reaction of 5b with lithium chloride in THF at 65 °C in 80% yield: H. Paulsen, K. Eberstein, and W. Koebernick, Tetrahedron Lett., 4377 (1974).
- 14) The hydrogenation of 3b with 5% Pd on charcoal giving 6-hydroxy-2-methyltetrahydropyran-3-one has been reported: Ref. 9b.
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